Name _____ Class_____Date_____

Skills Worksheet

Section Review

Development of the Atomic Theory

USING KEY TERMS

1. In your own words, write a definition for the term *atom*.

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The statements below are false. For each statement, replace the underlined term to make a true statement.

2. A <u>nucleus</u> is a particle with a negative electric charge.

_____

3. The <u>electron</u> is where most of an atom's mass is located.

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UNDERSTANDING KEY IDEAS

_____ 4. Which of the following scientists discovered that atoms contain electrons?
   a. Dalton
   b. Thomson
   c. Rutherford
   d. Bohr

5. What did Dalton do in developing his theory that Democritus did not do?

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6. What discovery demonstrated that atoms are mostly empty space?

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7. What refinements did Bohr make to Rutherford's proposed atomic theory?

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CRITICAL THINKING

8. **Making Comparisons** Compare the location of electrons in Bohr's theory with the location of electrons in the current atomic theory.

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9. **Analyzing Methods** How does the design of Rutherford's experiment show what he was trying to find out?

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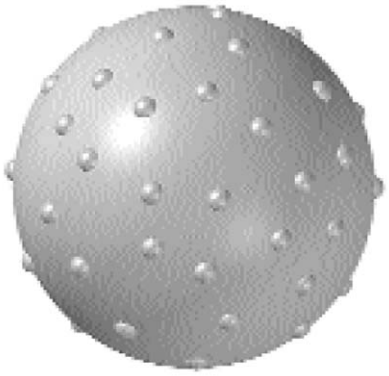
INTERPRETING GRAPHICS

10. What about the atomic model shown below was shown to be incorrect?



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